

Fluoride Salt-cooled High Temperature Reactors – Technology Status and Development Strategy

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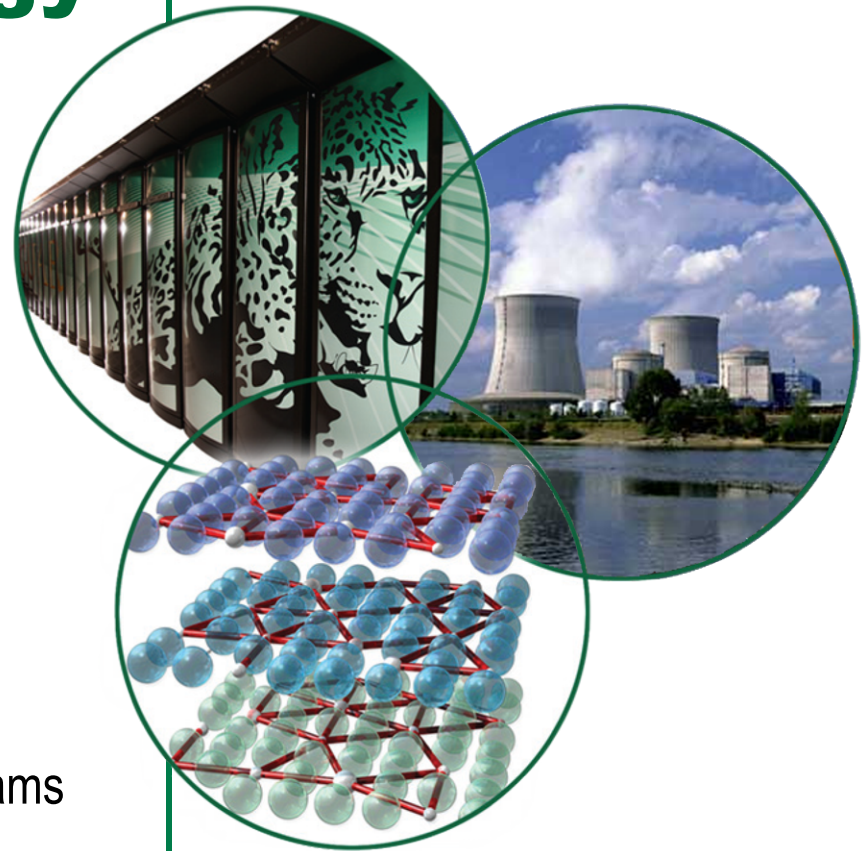
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Presentation overview

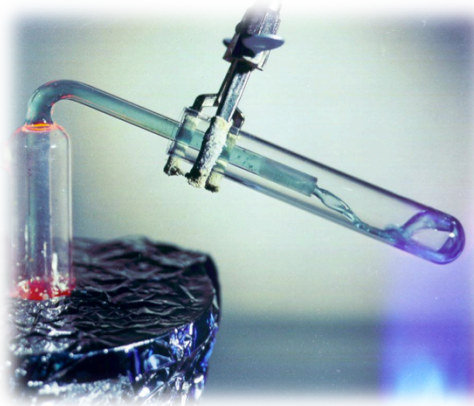
- **Fluoride salt**
- **Fluoride Salt-Cooled High Temperature Reactors (FHRs)**
- **FHR Technology Suite**
- **FHR Development Challenges**
- **FHR Development Strategy**

Presentation overview

- **Fluoride salt**

What the heck is “fluoride salt” ?

PERIOD	GROUP 1 IA	GROUP 2 IIA
1	1 1.0079 H HYDROGEN	
2	3 6.941 Li LITHIUM	4 9.012 Be BERYLLIUM
3	11 22.990 Na SODIUM	12 24.305 Mg MAGNESIUM
4	19 39.098 K POTASSIUM	20 40.078 Ca CALCIUM
5	37 85.468 Rb RUBIDIUM	38 87.62 Sr STRONTIUM
6	55 132.91 Cs CAESIUM	56 137.33 Ba BARIUM
7	87 (223) Fr FRANCIUM	88 (226) Ra RADIUM



Liquid and “Frozen” $2\text{LiF}\cdot\text{BeF}_2$ salt

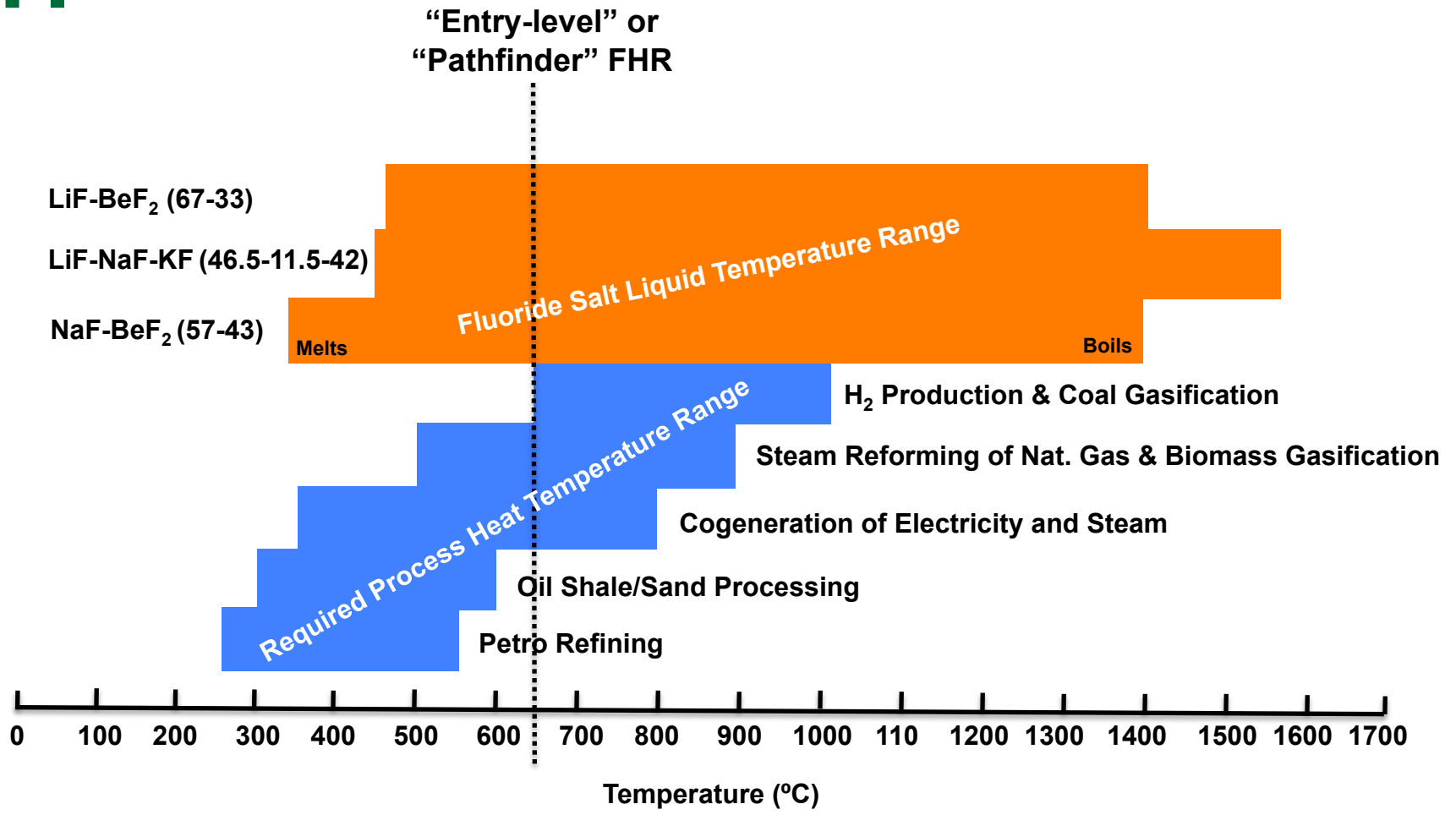
PERIOD	GROUP 17 VIIA	GROUP 18 VIIIA
1		2 4.0026 He HELIUM
2	9 18.998 F FLUORINE	10 20.180 Ne NEON
3	17 35.453 Cl CHLORINE	18 39.948 Ar ARGON
4	35 79.904 Br BROMINE	36 83.80 Kr KRYPTON
5	53 126.90 I IODINE	54 131.29 Xe XEON
6	(209) 85 (210) At ASTATINE	(222) 86 (222) Rn RADON

- “Fluoride salt” is a “halide salt”
- Halide salts are ionic compounds formed from the combination of a **halogen** and another element – commonly, but not exclusively, **alkali metals or alkaline earths**
- Examples: LiF , BeF_2 , KF , NaF , ZrF_4 , RbF , and mixtures of same

The potential benefits and challenges of FHRs stem from fundamental coolant properties

Coolant	T _{melt} (°C)	T _{boil} (°C)	Density (kg/m ³)	Specific Heat (kJ/kg °C)	Volumetric Heat Capacity (kJ/m ³ °C)	Thermal Conductivity (W/m °C)	Kinematic Viscosity (m ² /s) * 10 ⁶
Li₂BeF₄ (Flibe)	459	1430	1940	2.42	4670	1.0	2.9
59.5NaF-40.5ZrF₄	500	1290	3140	1.17	3670	0.49	2.6
26LiF-37NaF-37ZrF₄	436		2790	1.25	3500	0.53	
31LiF-31NaF-38BeF₂	315	1400	2000	2.04	4080	1.0	2.5
8NaF-92NaBF₄	385	700	1750	1.51	2640	0.5	0.5
Sodium	97.8	883	820	1.27	1040	62	0.12
56Na-44K	19	826	759	1.04	789	28.4	0.25
22Na-78K	-11	784	742	0.87	646	26.8	0.24
Lead	328	1750	10540	0.16	1686	16	0.13
44.5Pb-55.5Bi	98	881	10020	0.15	1503	13.9	0.12
Helium, 7.5 MPa			3.8	5.2	20	0.29	11
Water, 7.5 MPa	0	290	732	5.5	4040	0.56	0.13

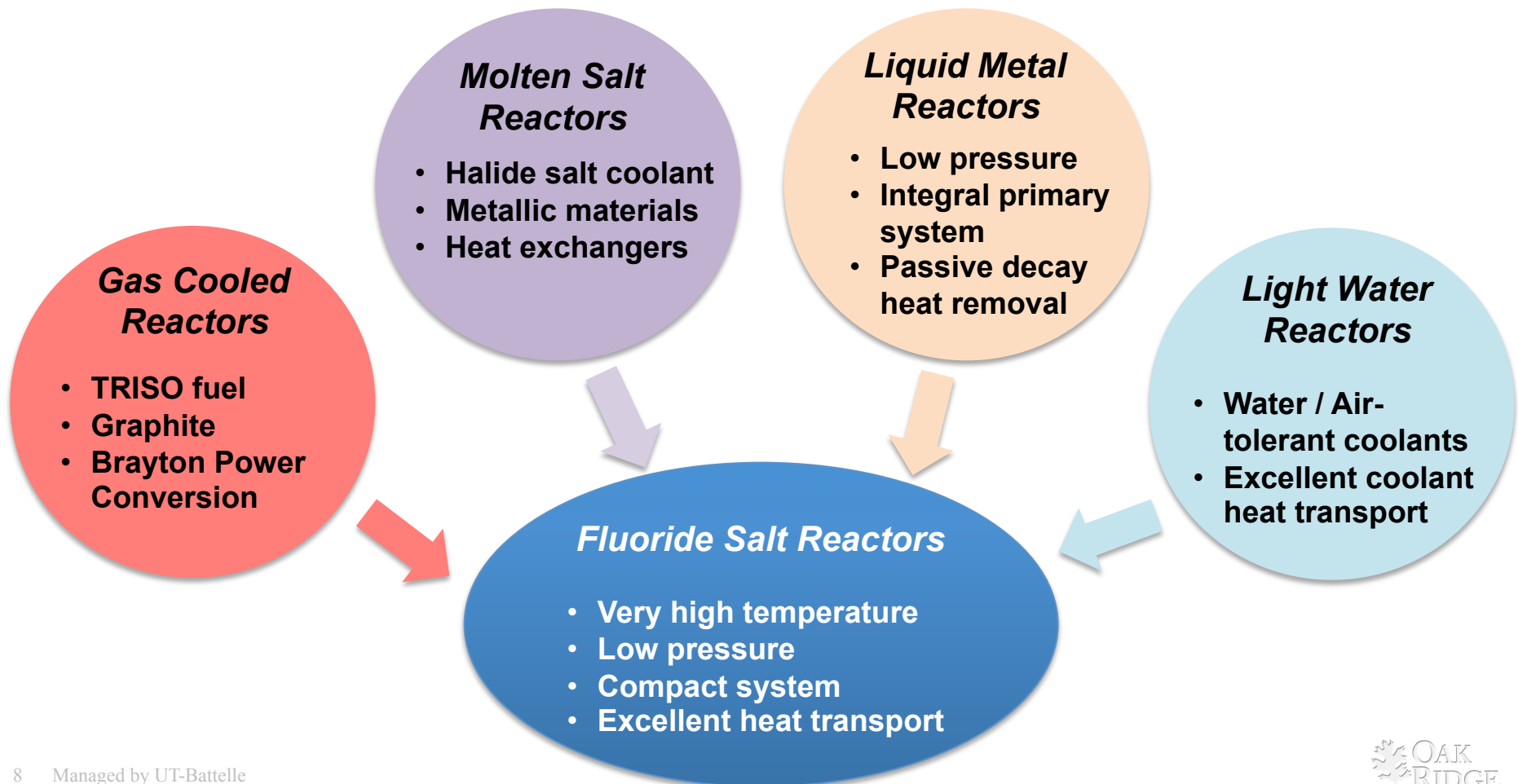
Potential FHR operating temperatures match many important process heat applications



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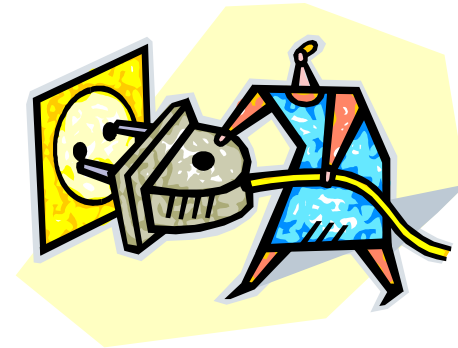
FHRs combine the best attributes of other reactor types to provide unique performance benefits



FHRs are promising candidates for traditional and non-traditional applications

- Electricity production

- Large centralized
- Small remote site



- High and Very High-Temperature Process Heat production

- Large centralized
- Small remote site

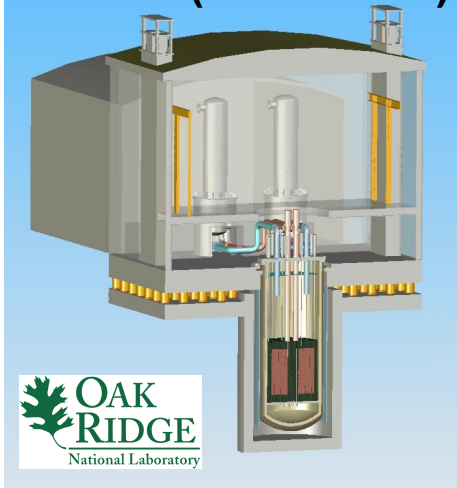


- Incremental energy demand growth scenarios

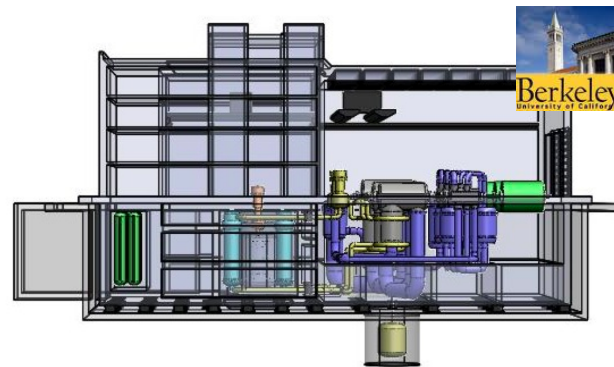
- Compact power applications

FHR concepts are being developed for diverse applications

AHTR* (1500 MWe)



PB-AHTR* (410 MWe)



SmAHTR* (125 MWt/50 Mwe)

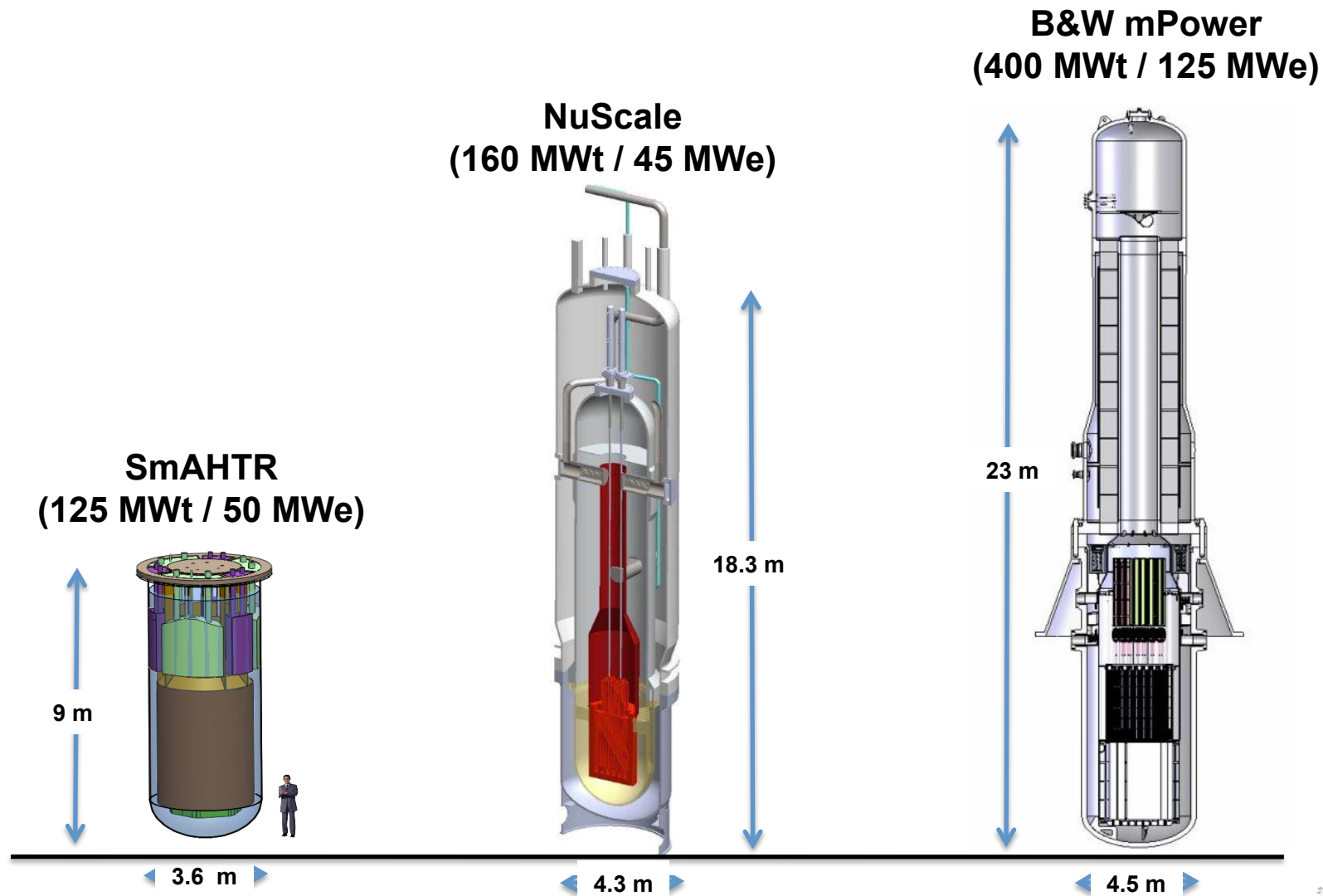


* AHTR = Advanced High Temperature Reactor





















PB-AHTR = Pebble Bed Advanced High Temperature Reactor

SmAHTR = Small Modular Advanced High Temperature Reactor

FHRs can be attractive Small Modular Reactors (SMRs)



FHRs incorporate many attractive attributes

Coolant (Reactor Concept)	High Working Temp ^a	High Volumetric Heat Capacity ^b	Low Primary Pressure ^c	Low Reactivity With Air & Water ^d	Coolant & Materials Cost
Water (PWR)					
Sodium (SFR)					
Helium (GCR)					
Salt (AHTR)					

^a High system working temperature desirable for high efficiency power conversion and process heat applications

^b High coolant volumetric heat capacity enables ~constant temperature heat addition / removal ($\eta_C = 1 - T_C/T_H \sim$ Carnot cycles), compact system architectures, and reduces pumping power requirements

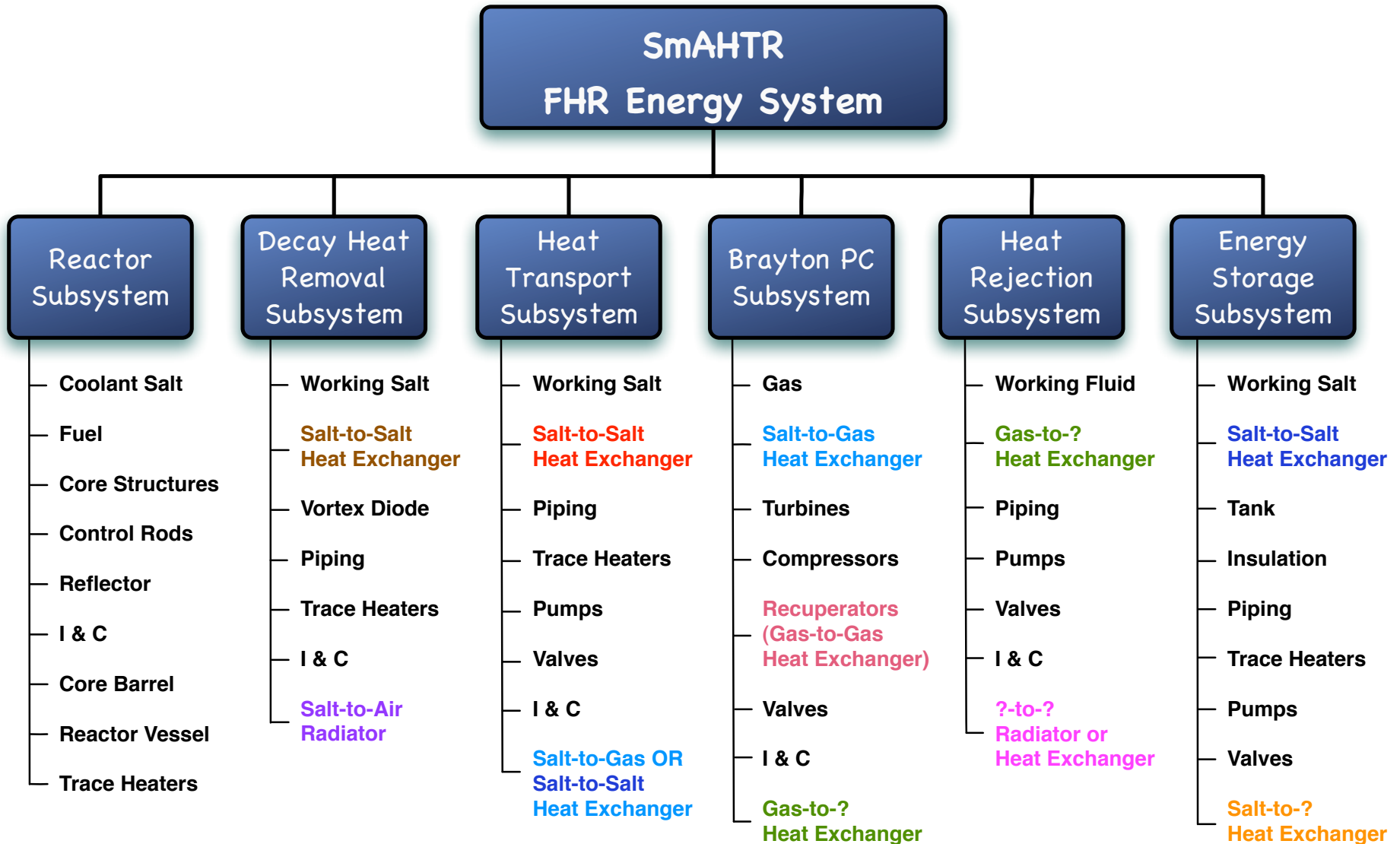
^c Low primary system pressure reduces cost of primary vessel and piping, and reduces energetics of pipe break accidents

^d Low reactivity with air and water reduces energetics of pipe break accidents

Presentation overview

- Fluoride salt
- Fluoride Salt-Cooled High Temperature Reactors (FHRs)
- **FHR Technology Suite**
- FHR Development Challenges
- FHR Development Strategy

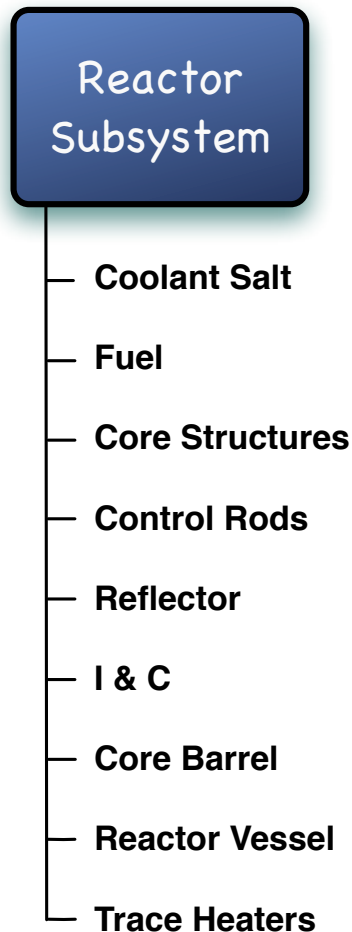
FHR critical technology maturity varies



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Reactor subsystem technology development focuses on materials



- Fluoride salt – LiF-BeF_2 is coolant of choice
 - Requires 99.993% ^7Li enrichment to maintain negative void coefficient and avoid unacceptable tritium production in reactor coolant
 - Has Be occupational health concerns
 - Expensive
- Fuel – NGNP / AGR TRISO fuel
 - Solid and annular pellets, plates, or pebbles are possible
- Core Structures
 - C-C and SiC-SiC composites
- Control Rods
 - C-C guides
 - Boron carbide or molybdenum hafnium carbide
- Reflector – Nuclear grade graphite
- Core Barrel – C-C composite
- Reactor Vessel – Alloy N

Li / Be / F compounds are very good (but expensive) coolants from nuclear perspective

Salt Constituent	Moderating Ratio	Short-lived Activation	Long-lived Activation
Lithium-7	Good	Very Good	Very Good
Beryllium	Very Good	Very Good	Good
Fluorine	Very Good	Very Good	Very Good
Sodium	Acceptable	Acceptable	Good
Potassium	Poor	Acceptable	Poor
Rubidium	Acceptable	Poor	Good
Zirconium	Good	Poor	Acceptable

- All fluoride salts are good coolants
- Lighter (low-Z) salts
 - Better heat transfer and nuclear performance
 - Have larger moderating ratios, yielding more negative (better) coolant void coefficients (FLiBe is best)
- Heavy (high-Z) salts
 - Lower heat capacities and thermal conductivities
 - More activation and transmutation products

Salt optimization (value engineering) for heat transport and storage applications is needed

Decay Heat Removal Subsystem

Heat Transport Subsystem

Energy Storage Subsystem

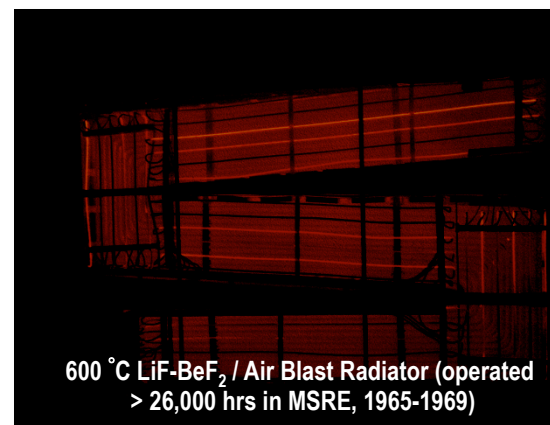
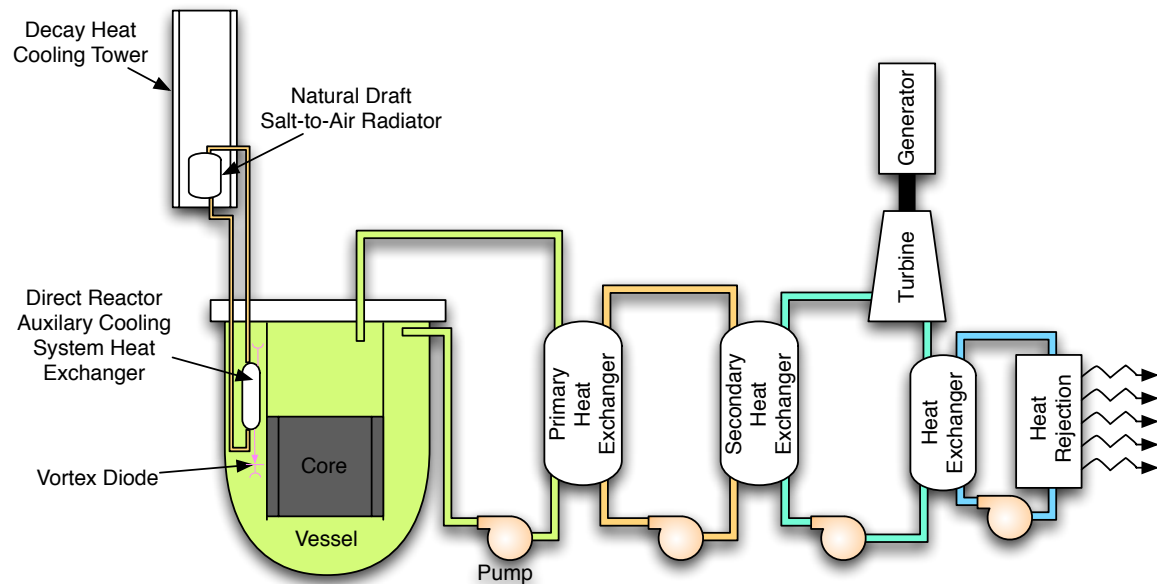
Salt	Formula Weight (g/mole)	Melting Point (°C)	900 °C Vapor Pressure (mm Hg)	ρ Density (g/cm ³)	PC_p Volumetric Heat Capacity (cal/cm ³ ·°C)	μ Viscosity (cP)	k Thermal Conductivity (W/m-K)
LiF-NaF-KF	41.3	454	~0.7	2.02	0.91	2.9	0.92
NaF-ZrF ₄	92.7	500	5	3.14	0.88	5.1	0.49
KF-ZrF ₄	103.9	390	1.2	2.80	0.70	<5.1	0.45
LiF-NaF-ZrF ₄	84.2	436	~5	2.92	0.86	6.9	0.53
LiCl-KCl	55.5	355	5.8	1.52	0.435	1.15	0.42
LiCl-RbCl	75.4	313	—	1.88	0.40	1.30	0.36
NaCl-MgCl ₂	73.7	445	<2.5	1.68	0.44	1.36	0.59
KCl-MgCl ₂	81.4	426	<2.0	1.66	0.46	1.40	0.40
NaF-NaBF ₄	104.4	385	9500	1.75	0.63	0.90	0.40
KF-KBF ₄	109.0	460	100	1.70	0.53	0.90	0.38
RbF-RbF ₄	151.3	442	<100	2.21	0.48	0.90	0.28

Very little R&D to optimize fluoride salt properties and salt production processes has been conducted since 1960's

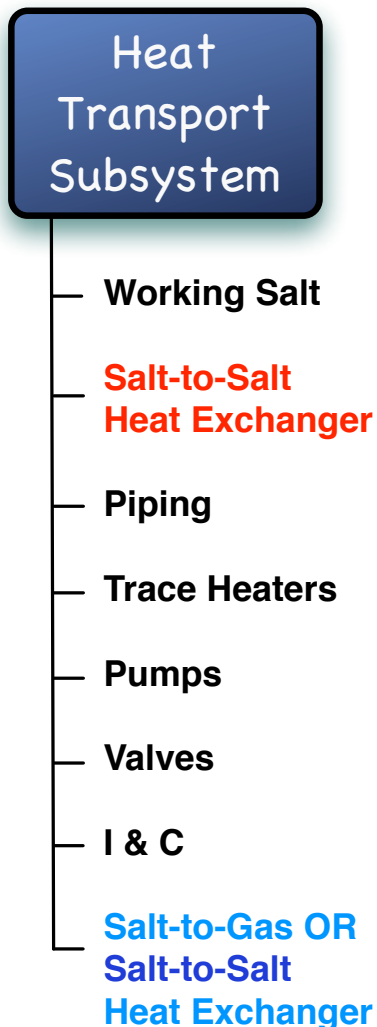
Heat exchangers and vortex diodes are principal passive decay heat removal subsystem technology development needs

Decay Heat Removal Subsystem

- Working Salt
- Salt-to-Salt Heat Exchanger
- Vortex Diode
- Piping
- Trace Heaters
- I & C
- Salt-to-Air Radiator



Heat exchangers, pumps, and valves are primary heat transport subsystem technology development challenges

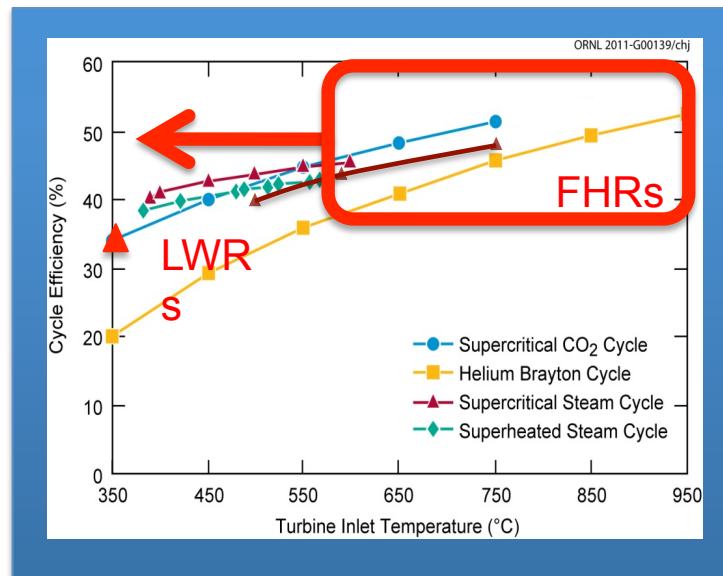


- **Pumps**
 - Long-shafted centrifugal pumps
 - Ceramic bearings and dry gas seals
 - Ti-modified Alloy N material
- **Piping**
 - Alloy N for temperatures < 700 °C
 - C-C composites for higher temperatures
- **Valves**
 - No experience base for active valving
 - “Freeze plugs” used in MSRE
 - Salt-compatible valve seats and seals
- **Trace Heaters & Insulation**
 - No major challenges

Optimal Brayton power conversion technology is not obvious...

Brayton PC Subsystem

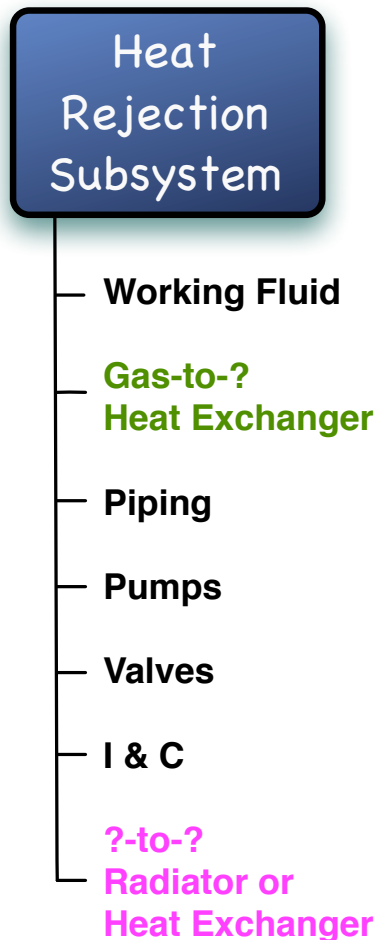
- Gas
- Salt-to-Gas Heat Exchanger
- Turbines
- Compressors
- Recuperators (Gas-to-Gas Heat Exchanger)
- Valves
- I & C
- Gas-to-? Heat Exchanger



Data from: W. S. Jeong et al., *Potential Improvements of Supercritical CO₂*, Korea Advanced Institute of Science and Technology; and I. Satyanarayana et al., *Second Law Analysis of Super Critical Cycle*, International Journal of Engineering (IJE), Volume (4): Issue (1)

- Closed-cycle Brayton:
 - He: High-temp capability, bulky, complicated
 - Super-critical CO₂, compact, temp-limited
- Open direct-cycle air
 - Very high temperature capability, simple
- Supercritical steam Rankine is mature

Heat rejection subsystem technology needs are coupled to power conversion technology selection

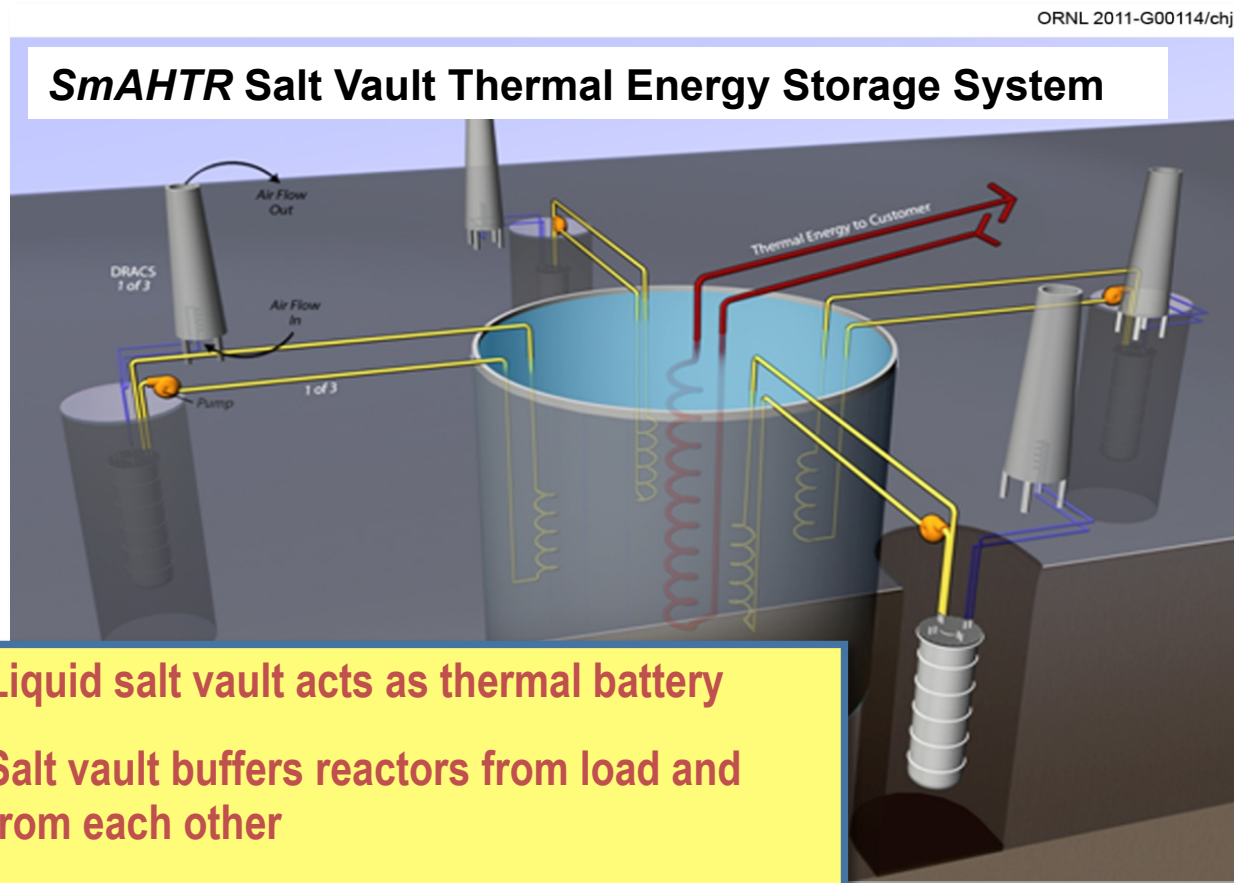


- Almost no FHR-specific work done to date.
- Coupled to power conversion selection
- “Dry” heat rejection to air should be attractive due to high temperatures
- Hybrid wet/dry systems may be 1st step
- Opportunity to leverage concentrating solar power plant technologies?

Feasibility of FHR thermal energy storage system hinges on cost of salt

Energy Storage Subsystem

- Working Salt
- Salt-to-Salt Heat Exchanger
- Tank
- Insulation
- Piping
- Trace Heaters
- Pumps
- Valves
- Salt-to-? Heat Exchanger



- Liquid salt vault acts as thermal battery
- Salt vault buffers reactors from load and from each other
- Salt selection and salt vault size can be optimized for differing applications
 - 125 MWt-hr storage @ 500 –600 °C requires ~ 13 meter cubic salt tank

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Near-term technical feasibility of FHRs is dominated by four considerations

- “Cost” of salt coolants (especially LiF-BeF₂) and heat-transport working salts
 - Li-7 isotopic purity is major cost driver
 - Be ES&H considerations
 - Bulk heat transport/storage compounds
- Tritium management technology for systems employing Li-based reactor coolants
- Alloy N code qualification (in-vessel) *and supply chain*
- Heat exchanger and radiator technology
 - High temperature
 - Salt-to-salt, salt-to-gas, salt-to-air
 - Feasibility of high pressure differential designs

An integrated FHR development strategy would focus on

- System concept development
- Base technology development
 - Optimized salts
 - Salt cleanup and tritium management
 - TRISO fuels (leverage on-going gas reactor fuels work)
 - High-temperature materials (especially advanced composites)
- Component development
 - Heat exchangers (salt-to-salt, salt-to-gas)
 - Pumps
 - Valves
 - Fluidic diodes
- High temperature power conversion technology
- Infrastructure
 - Salt synthesis laboratories
 - Electrically heated loops for materials, component, power conversion, and heat rejection technology development
- Small demonstration FHR

Materials R&D will pace evolution to higher operating temperatures

System Element	@ 700 °C	@ 850 °C	@ 1000 °C
Graphite Internals	Toyo Tanso IG110 or 430	Toyo Tanso IG110 or 430	Toyo Tanso IG110 or 430
Reactor Vessel	Alloy N	<ul style="list-style-type: none"> •Ni-weld overlay on 800H •Insulated low-alloy steel •New Ni-based alloy 	<ul style="list-style-type: none"> • Interior-insulated low-alloy steel
Core barrel & other internals	Alloy N	<ul style="list-style-type: none"> •C-C composite •New Ni-based alloy 	<ul style="list-style-type: none"> •C-C composite •SiC-SiC composite •New refractory metal
Control rods and internal drives	<ul style="list-style-type: none"> •Alloy N •C-C composites •Nb-1Zr or MHC 	<ul style="list-style-type: none"> •C-C composites •Nb-1Zr •MHC 	<ul style="list-style-type: none"> •C-C composites •Nb-1Zr •MHC
PHX & DRACS	Alloy N	<ul style="list-style-type: none"> •New Ni-based alloy •Double-sided Ni cladding on 617 or 230 	<ul style="list-style-type: none"> •C-C composite •SiC-SiC composite •Monolithic SiC
Secondary (salt-to-gas) HX	Coaxial extruded 800H tubes with Ni-based layer	<ul style="list-style-type: none"> •New Ni-based alloy •Coaxial extruded 800H tubes with Ni-based layer 	?

Summary

- FHRs are a new class of high- to very-high-temperature reactor that leverages best features of traditional reactors
- FHR designs target :
 - process heat production and electricity generation
 - ease of transport and deployment
 - long-term evolvability to higher efficiency electric generation and higher temperature process heat applications
- A 650°C FHR is a reasonable “pathfinder” objective
- Current technology status benefits from molten salt reactor, gas-cooled, and liquid metal-cooled reactor technology development over many decades
- Key challenges are salt economics, high-temperature materials, and high-temperature heat exchanger technologies
- Reason for technical optimism
- Promise is significant